

Guido Kickelbick,^a Richard Hoogenboom^b and Ulrich S. Schubert^{b*}^aVienna University of Technology, Institute of Materials Chemistry, Getreidemarkt 9/165, A-1060 Wien, Austria, and ^bLaboratory of Macromolecular Chemistry and Nanoscience, Eindhoven University of Technology, PO Box 513, 5600 MB Eindhoven, The Netherlands

Correspondence e-mail: u.s.schubert@tue.nl

Key indicators

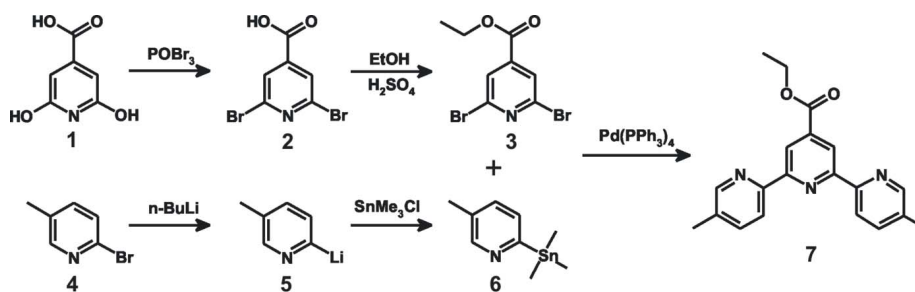
Single-crystal X-ray study
 $T = 173$ K
Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å
 R factor = 0.049
 wR factor = 0.149
Data-to-parameter ratio = 18.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Ethyl 5,5''-dimethyl-2,2';6',2''-terpyridine-4'-carboxylate

The title compound, $\text{C}_{20}\text{H}_{19}\text{N}_3\text{O}_2$, was synthesized by Stille coupling of ethyl 2,6-dibromoisonicotinate and 2-trimethylstannyl-5-methylpyridine. The three pyridine rings in the molecule are coplanar and the crystal packing reveals π - π stacking interactions between these conjugated aromatic rings with a distance between the mean planes of 3.494 Å.

Comment

2,2':6',2''-Terpyridine is a well known chelating ligand for a wide variety of transition metal ions (Hofmeier & Schubert, 2004). This supramolecular binding motif has been applied for the assembly of, for example, metallodendrimers (Constable, 1997) and metallocsupramolecular polymers (Schubert & Eschbaumer, 2002; Andres & Schubert, 2004). The synthesis of functionalized 2,2':6',2''-terpyridines has been reviewed in recent years by Cargill Thompson (1997), Fallahpour (2003) and Heller & Schubert (2003). The crystal structure of the unsubstituted 2,2':6',2''-terpyridine was reported by Bessel *et al.* (1992). Subsequently, crystal structures of 4'-substituted 2,2':6',2''-terpyridines, such as 4'-vinyl-2,2':6',2''-terpyridine (Liu *et al.*, 2000), 4'-butoxy- and 4'-dodecyloxy-2,2':6',2''-terpyridine (Andres *et al.*, 2003), and 4'-(5-isocyanatopentyl-oxy)-2,2':6',2''-terpyridine (Hoogenboom *et al.*, 2004), have been reported. However, only a few crystal structures of 6,6''-substituted 2,2':6',2''-terpyridines have been described in the literature, namely 6-[(1*S*)-endo]-(-)-bornyloxy-2,2':6',2''-terpyridine, 6-[(1*S*)-endo]-(-)-bornyloxy-6''-methyl-2,2':6',2''-terpyridine (Baum *et al.*, 2000), 4,4-difluoro-8-(6''-methyl-2',2'':6'',2'''-terpyridin-6'-yl)-1,3,5,7-tetramethyl-2,4-diethyl-4-bora-3a,4a-diaza-*s*-indacene (Goze *et al.*, 2003), 6,6''-bis(trimethylsilylethynyl)-2,2':6',2''-terpyridine and 6,6''-bis(trimethylsilylethynyl)-4'-phenyl-2,2':6',2''-terpyridine (Khan *et al.*, 2002). To the best of our knowledge, the crystal structure of this latter compound is the only structure reported for a 2,2':6',2''-terpyridine with substituents on all three rings.



We report here the crystal structure of another 2,2':6',2''-terpyridine with three substituents, namely the title

Received 10 November 2005

Accepted 21 November 2005

Online 26 November 2005

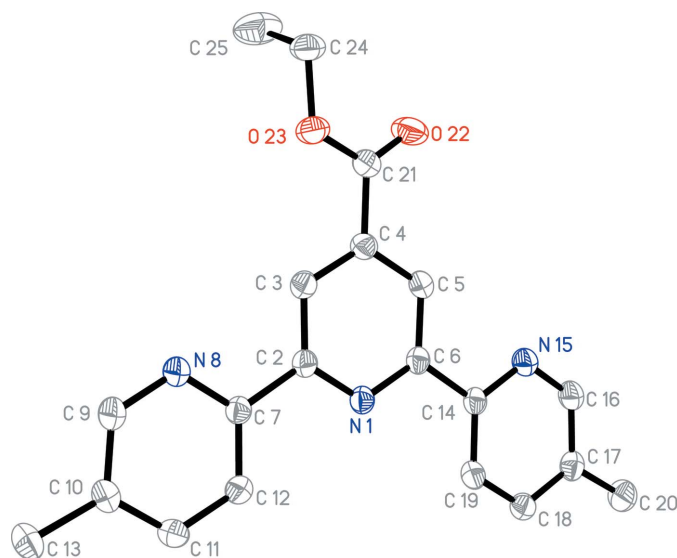


Figure 1
The structure of (7), with displacement ellipsoids shown at the 50% probability level. For clarity, H atoms have been omitted.

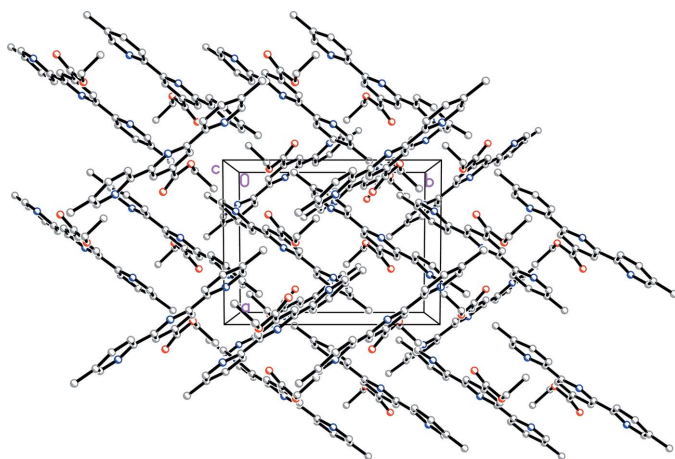


Figure 2
Projection of the structure along [001]. For clarity, H atoms have been omitted.

compound, (7). The molecular structure of (7) is shown in Fig. 1. The three rings of the terpyridine unit, N1/C2–C6, C7/N8/C9–C12 and C14/N15/C16–C19, are coplanar. The plane through the terpyridine unit makes an angle to the carboxylic ester group (C21/O22/O23) of 10.99 (4)°. All bond lengths and angles can be regarded as normal. The crystal packing reveals π – π stacking interactions between the conjugated aromatic rings in the structure, with a distance between the mean planes of 3.494 Å (Fig. 2).

Experimental

Compound (7) was synthesized starting from 2,6-dihydroxyisonicotinic acid, (1), and 2-bromo-5-methylpyridine, (4), following a modified literature procedure (Fallahpour, 2000; Heller & Schubert, 2002). Compound (1) was converted into 2,6-dibromoisonicotinic acid ethyl ester, (3), *via* 2,6-dibromoisonicotinic acid, (2). 2-Bromo-5-methylpyridine was lithiated, resulting in (5), and converted into 2-

trimethylstannyl-5-methylpyridine, (6), by the addition of trimethylstannylchloride (in the literature method 2-tributylstannyl-5-methylpyridine was used). Compounds (3) and (6) were coupled *via* a Pd(PPh₃)₄-catalysed Stille coupling, resulting in the title compound (7), which crystallized as single crystals by slow evaporation of a CDCl₃ solution. The reaction scheme of the synthesis is shown above.

Crystal data

C₂₀H₁₉N₃O₂
M_r = 333.38
 Monoclinic, *P*2₁/*n*
a = 9.0892 (5) Å
b = 11.8780 (6) Å
c = 15.8131 (8) Å
 β = 97.975 (1)°
V = 1690.70 (15) Å³
Z = 4

D_x = 1.310 Mg m⁻³
 Mo K α radiation
 Cell parameters from 5732 reflections
 θ = 4.9–56.5°
 μ = 0.09 mm⁻¹
T = 173 (2) K
 Irregular fragment, colourless
 0.56 × 0.41 × 0.26 mm

Data collection

Bruker SMART CCD area-detector diffractometer
 ω scans
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.953, *T_{max}* = 0.978
 11535 measured reflections

4169 independent reflections
 3408 reflections with *I* > 2 σ (*I*)
R_{int} = 0.028
 θ_{max} = 28.3°
h = –12 → 12
k = –15 → 15
l = –11 → 21

Refinement

Refinement on *F*²
R[*F*² > 2 σ (*F*²)] = 0.049
wR(*F*²) = 0.149
S = 1.06
 4169 reflections
 229 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0991P)^2 + 0.1562P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} = 0.006$
 $\Delta\rho_{max} = 0.33 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{min} = -0.37 \text{ e } \text{Å}^{-3}$

Table 1

Selected geometric parameters (Å, °).

C10–C13	1.5066 (15)	C21–O23	1.3365 (14)
C17–C20	1.5070 (16)	O23–C24	1.4580 (14)
C21–O22	1.2003 (14)		
N1–C2–C7	116.99 (9)	O23–C21–C4	111.74 (9)
N1–C6–C14	116.79 (9)	C21–O23–C24	117.03 (9)
O22–C21–O23	124.52 (11)	O23–C24–C25	110.32 (11)
O22–C21–C4	123.74 (11)		
N1–C2–C7–N8	–178.07 (9)	O22–C21–O23–C24	2.06 (18)
N1–C6–C14–N15	176.78 (9)	C21–O23–C24–C25	91.01 (14)
C3–C4–C21–O22	–167.76 (12)		

H atoms were located in difference Fourier maps and refined with a riding model, with C–H distances of 0.95 (aromatic H), 0.98 (methyl H) and 0.99 Å (methylene H), and with *U*_{iso}(H) = 1.2*U*_{eq}(C) for aromatic and methylene H, or 1.5*U*_{eq}(C) for methyl H.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1998); software used to prepare material for publication: *SHELXTL*.

Guido Kickelbick thanks the Fonds zur Förderung der wissenschaftlichen Forschung (FWF), Austria, for financial support. Richard Hoogenboom and Ulrich S. Schubert thank the Dutch Scientific Organization (NWO) and the Fonds der Chemischen Industrie for financial support.

References

- Andres, P. R., Lunkwitz, R., Pabst, G. R., Böhn, K., Wouters, D., Schmatloch, S. & Schubert, U. S. (2003). *Eur. J. Org. Chem.* pp. 3769–3776.
- Andres, P. R. & Schubert, U. S. (2004). *Adv. Mater.* **16**, 1043–1068.
- Baum, G., Constable, E. C., Fenske, D., Housecroft, C. E., Kulke, T., Neuburger, M. & Zehnder, M. (2000). *J. Chem. Soc. Dalton Trans.* pp. 945–959.
- Bessel, C. A., See, R. F., Jameson, D. L., Churchill, M. R. & Takeuchi, K. J. (1992). *J. Chem. Soc. Dalton Trans.* pp. 3223–3228.
- Bruker (1997). *SMART* and *SAINT*. Bruker AXS GmbH, Karlsruhe, Germany.
- Bruker (1998). *SHELXTL*. Version 5.1. Bruker AXS GmbH, Karlsruhe, Germany.
- Cargill Thompson, A. M. W. (1997). *Coord. Chem. Rev.* **160**, 1–52.
- Constable, E. C. (1997). *Chem. Commun.* pp. 1073–1080.
- Fallahpour, R.-A. (2000). *Synthesis*, pp. 1138–1142.
- Fallahpour, R.-A. (2003). *Synthesis*, pp. 155–184.
- Goze, C., Ulrich, G., Charbonnière, L., Cesario, M., Prange, T. & Ziessel, R. (2003). *Chem. Eur. J.* **9**, 3748–3755.
- Heller, M. & Schubert, U. S. (2002). *J. Org. Chem.* **67**, 8269–8272.
- Heller, M. & Schubert, U. S. (2003). *Eur. J. Org. Chem.* pp. 947–961.
- Hofmeier, H. & Schubert, U. S. (2004). *Chem. Soc. Rev.* **33**, 373–399.
- Hoogenboom, R., Andres, P. R., Kickelbick, G. & Schubert, U. S. (2004). *Synlett*, **10**, 1779–1783.
- Khan, M. S., Al-Mandhary, M. R. A., Al-Suti, M. K., Hisahm, A. K., Raithby, P. R., Ahrens, B., Mahon, M. F., Male, L., Marseglia, E. A., Tedesco, E., Friend, R. H., Köhler, A., Feeder, N. & Teat, S. J. (2002). *J. Chem. Soc. Dalton Trans.* pp. 1358–1368.
- Liu, X., Kilner, C. A., Thornton-Pett, M. & Halcrow, M. A. (2000). *Acta Cryst.* **C56**, 1142–1143.
- Schubert, U. S. & Eschbaumer, C. (2002). *Angew. Chem. Int. Ed.* **41**, 2892–2926.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.